

Name_____

Chemistry I: Chemical Nomenclature: **WS 2012-10A**

Period _____ Date_____

Write the correct formula for:

1) silver carbonate		11) tin(IV) chlorite	
2) potassium phosphate		12) iron(II) phosphate	
3) aluminum hydroxide		13) tin(II) carbonate	
4) sodium hydrogen carbonate		14) ferric acetate	
5) calcium acetate		15) lead(II) chromate	
6) potassium permanganate		16) copper(I) sulfite	
7) calcium perchlorate		17) stannous dichromate	
8) lithium carbonate		18) iron(III) nitrate	
9) magnesium hydrogen sulfite		19) ferric sulfate	
10) sodium hypochlorite		20) ferrous hydroxide	

21) potassium perchlorate		28) plumbous cyanide	
22) lead(II) phosphate		29) silver phosphate	
23) aluminum sulfate		30) cobalt(III) nitrite	
24) iron(II) bicarbonate		and two special additions:	
25) barium iodate		31) ammonium sulfate	
26) tin(II) hydrogen sulfide		32) ammonium nitrate	
27) magnesium phosphate		31 & 32 are double polyatomic ions	